



**Conversations on Chemistry; In Which the
Elements of That Science Are Familiarly
Explained, and Illustrated by Experiments, and 38
Engravings on Wood**

Mrs Marcet

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This historic book may have numerous typos and missing text. Purchasers can download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1829 edition. Excerpt: ...Mr: B. Undoubtedly; when oxygen loses its gaseous form, in order to combine with a solid body, it becomes condensed, and the caloric evolved produces the elevation of temperature. (The specific gravity of bodies is at the same time altered by chemical combination; for in consequence of a change of capacity for heat, a change of density must be produced. Caroline. That was the case with the sulphuric acid and water, which, by being mixed together, gave out a great deal of heat, and increased in density. 718. How do acids ending in ic differ from those ending in ous? 719. What is the fourth law of chemical attraction? 720. From what does the change of temperature arise? 721. What is an instance of increase of temperature from chemical union? 722. Is the specific gravity of bodies affected by chemical combination: Mrs. B. The fifth law of chemical attraction is, that the properties which characterize bodies, when separate, are altered or destroyed by their combination? Caroline. Certainly? What, for instance, can be so different from water, as the hydrogen and oxygen gases? Emily. Or what more unlike sulphate of iron, than iron or sulphuric acid? Mrs. B. Every chemical combination is an illustration of this rule. But let us proceed---The sixth law is, that the force of chemical affinity between the constituents of a body, is estimated by that which is required for their separation. This force is not always proportional to the facility with which bodies unite; for manganese, for instance, which you know, is so much disposed to unite with oxygen, that it is never found in a metallic state, yields it more easily than any other metal. Emily. But, Mrs. B., you speak of estimating the force of attraction between bodies, by the force required to separate...

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